

1 **Nitrogen accumulation and partitioning in High Arctic tundra ecosystem**
2 **from extreme atmospheric N deposition events**

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16 Running Headline: Extreme N deposition event fate in High Arctic Tundra

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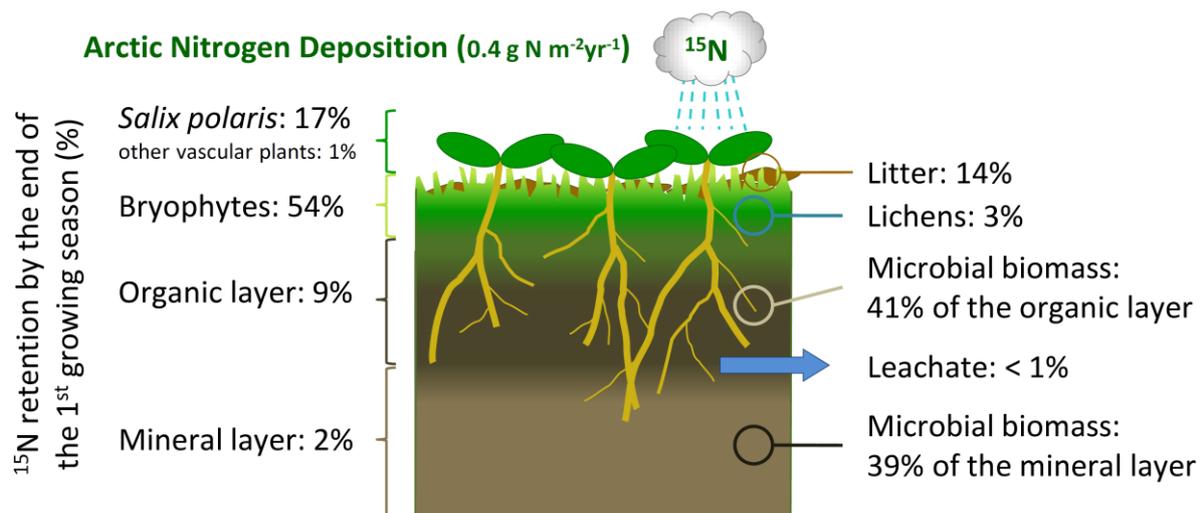
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27 Graphical abstract



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30 Highlights

- 31 • High Arctic tundra demonstrated a very high (90–95%) N pollutant retention capacity.
- 32 • Non-vascular plants and soil microbial biomass were important short-term N sinks.
- 33 • Vascular plants and soil showed capacity to be efficient longer-term N sinks.
- 34 • Deposition rich in nitrate can alter almost all ecosystem N pools.
- 35 • Extreme N depositions may already be contributing to the N enrichment of tundra.

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38 Abstract

39 Arctic ecosystems are threatened by pollution from recently detected extreme
40 atmospheric nitrogen (N) deposition events in which up to 90% of the annual N deposition
41 can occur in just a few days. We undertook the first assessment of the fate of N from extreme
42 deposition in High Arctic tundra and are presenting the results from the whole ecosystem ^{15}N
43 labelling experiment. In 2010, we simulated N depositions at rates of 0, 0.04, 0.4 and 1.2 g N
44 $\text{m}^{-2} \text{ yr}^{-1}$, applied as $^{15}\text{NH}_4^{15}\text{NO}_3$ in Svalbard (79°N), during the summer. Separate

45 applications of $^{15}\text{NO}_3^-$ and $^{15}\text{NH}_4^+$ were also made to determine the importance of N form in
46 their retention.

47 More than 95% of the total ^{15}N applied was recovered after one growing season
48 (~90% after two), demonstrating a considerable capacity of Arctic tundra to retain N from
49 these deposition events. Important sinks for the deposited N, regardless of its application rate
50 or form, were non-vascular plants N vascular plants N organic soil N litter N mineral soil,
51 suggesting that non-vascular plants could be the primary component of this ecosystem to
52 undergo measurable changes due to N enrichment from extreme deposition events.
53 Substantial retention of N by soil microbial biomass (70% and 39% of ^{15}N in organic and
54 mineral horizon, respectively) during the initial partitioning demonstrated their capacity to act
55 as effective buffers for N leaching. Between the two N forms, vascular plants (*Salix polaris*)
56 in particular showed difference in their N recovery, incorporating four times greater $^{15}\text{NO}_3^-$
57 than $^{15}\text{NH}_4^+$, suggesting deposition rich in nitrate will impact them more. Overall, these
58 findings show that despite the deposition rates being extreme in statistical terms, biologically
59 they do not exceed the capacity of tundra to sequester pollutant N during the growing season.
60 Therefore, current and future extreme events may represent a major source of eutrophication.

61

62 **Key-words**

63 Extreme nitrogen deposition, arctic tundra N pools, N immobilization, ^{15}N tracer, plant–soil
64 interactions, ecosystem N dynamics

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70 **1. Introduction**

71 Atmospheric nitrogen deposition is one of the top three threats to global biodiversity
72 (Sala et al., 2000; Phoenix et al., 2006, 2012). Since the industrial revolution, there has been a
73 marked increase in nitrogen (N) deposition across many regions of the world, including
74 supposedly pristine remote locations such as the Arctic (Jónsdóttir et al., 1995; Forsius et al.,
75 2010).

76 Despite this, arctic ecosystems still typically receive relatively low rates of
77 atmospheric N deposition (ranging from $<0.1 \text{ g N m}^{-2} \text{ yr}^{-1}$ in Svalbard and other arctic
78 regions to $\sim 1 \text{ g N m}^{-2} \text{ yr}^{-1}$ in the Taymyr Peninsular in Russia and parts of northern Alaska;
79 Woodin, 1997; Fischer et al., 1998; Simoões and Zagorodnov, 2001; Kühnel et al., 2011).
80 Therefore, even relatively modest increases in current and future N inputs may represent a
81 significant additional supply of N. This is of particular concern given that arctic ecosystems
82 are sensitive to increased N supply (Gordon et al., 2001; Arens et al., 2008; Street et al.,
83 2015) due to the inherent low N availability in these nutrient-limited systems (Shaver and
84 Chapin, 1980; Henry et al., 1986; Chapin et al., 1995).

85 While chronic rates of N deposition in the Arctic are low, a greater threat may arise
86 from extreme N deposition events that have been discovered relatively recently (Hodson et
87 al., 2005, 2010; Kühnel et al., 2011, 2013). The extremeness of the atmospheric N deposition
88 event refers to its high intensity (or concentration) as well as its low frequency i.e. a
89 statistically rare occurrence (Jentsch, 2006; Kühnel et al., 2011, 2013). Such extreme events
90 result from polluted air masses that arise from industrialised countries at lower latitudes and
91 are transported to high latitude regions with minimal dispersal. As an example, a single
92 extreme event was observed to supply 40% of the total annual deposition in just one week
93 (Hodson et al., 2005), and recent research suggests that despite their low frequency, because
94 of the large quantities of N deposited annual atmospheric N deposition in High Arctic

95 Svalbard can be dominated by such episodic events with these forming 10–90% of the annual
96 atmospheric N input (Hodson et al., 2005; Kühnel et al., 2011, 2013). It is of further concern
97 that increasing cyclonic activity over the North Atlantic and predicted increases in
98 precipitation over the Arctic may lead to more extreme N deposition events (Klonecki et al.,
99 2003; Kühnel et al., 2011). Furthermore, the changing arctic climate (Førland et al., 1997;
100 Cassano et al., 2006) and intensification of shipping activities (Serreze et al., 2007; Peters et
101 al., 2011) may lead especially to particularly greater increases in deposition events during the
102 summer, the time of year when arctic ecosystems will have the greatest biological capacity to
103 sequester the N, potentially exacerbating N enrichment impacts. To date, however, the fate
104 and impacts of N from these extreme deposition events remains unknown, despite the
105 potential for these to affect arctic ecosystems by loading a large proportion of the annual
106 atmospheric N input in just a few days. The effects of such extreme depositions will depend
107 in part on the amount of N retained in the system and its partitioning among different
108 compartments of the ecosystem(plant, soil, microbial, leachate pools) cross time.
109 Quantification of the fate of N may also allow for a better understanding of the mechanisms
110 underlying shifts in species composition of an ecosystem.

111 Many nitrogen manipulation experiments in tundra have used NPK fertiliser additions
112 rather than simulating atmospheric N deposition (e.g., Shaver and Chapin, 1995; Robinson et
113 al., 1998; Schmidt et al., 2000) to study the effects of N availability. Other N addition studies
114 have either used large N applications (ranging from 5 to 25 g m⁻² yr⁻¹) or investigated
115 chronic N loading but to date have not simulated extreme atmospheric N deposition events
116 (e.g., Gordon et al., 2001; Madan et al., 2007; Arens et al., 2008).While a limited number of
117 studies have focused on the fate of atmospheric N deposition in the Arctic, these studied
118 mainly snowpack N inputs during spring melt (Bilbrough et al., 2000; Tye et al., 2005;
119 Templer et al., 2012). Moreover, none of the studies have demonstrated a comprehensive

120 understanding of atmospheric N partitioning and retention of different N forms (e.g.
121 $\text{NH}_4^+\text{NO}_3^-$, only- NH_4^+ and only- NO_3^-) within different ecosystem pools.

122 With these concerns in mind, we used a field simulation of extreme deposition events,
123 with ^{15}N (applied over 4 days in summer at 0, 0.04, 0.4 and 1.2 $\text{g N m}^{-2} \text{yr}^{-1}$) to determine the
124 fate (accumulation and partitioning) of acutely deposited N over two growing seasons. Our
125 objectives were (1) to identify major sinks for atmospheric N, deposited in extreme events,
126 and therefore, determining the most sensitive component of the ecosystem to be affected by
127 such depositions, (2) to determine the fate of different species of reactive N (NO_3^- and NH_4^+)
128 in different ecosystem compartments, and therefore, investigating if there is any preference
129 for either of the N forms by any of the compartments of the ecosystem as well as the form of
130 nitrogen that could impact this ecosystem more. Finally, we (3) analysed the capacity of
131 different doses and forms of ^{15}N in enriching the total N pool of the different ecosystem
132 compartments to understand the eutrophication potential of the deposited N. We report results
133 from the whole ecosystem for both the initial partitioning and recovery of ^{15}N label in
134 different pools: tundra plants, soil, litter, microbial biomass and leachate as well as the
135 cycling of ^{15}N over the short (7–21 days) and medium terms (388 days).

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137 **2. Material and methods**

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139 **2.1. Site description**

140 The field site was located in the High Arctic on Svalbard, at Leirhaugen Kolhaugen
141 ($78^\circ55'231''\text{N}$; $11^\circ49'819''\text{E}$), 25 m above mean sea level on the Brøggerhalvøya peninsula, 2
142 km southwest of Ny-Ålesund. The mean annual temperature at Ny-Ålesund is -5.2°C , with a
143 summer (July–August) average of $+5^\circ\text{C}$. The vegetation was dominated by bryophytes
144 (~40%; mainly: *Sanionia uncinata*, *Ptilidium ciliare*, *Dicranum laevidens*, *Dicranum*
145 *spadiceum* and *Oncophorus wahlenbergii*) and *Salix polaris* (a deciduous dwarf shrub:

146 ~30%) and lichens (~16%) with a thin layer of litter (mainly Salix) on the surface underlain
147 by an organic (O) layer (~5 cm) over a mineral (A) layer. Further details on climatic
148 conditions during the experiment, soils and vegetation C and N stocks are provided in
149 Appendix A (Tables A.1 & A.2).

150

151 **2.2. Experimental design**

152 In July 2009, 25 plots (1.5 m × 1.5 m) were established in approximately a 600 m²
153 area of tundra (see Blaud et al., 2015 and Appendix A for details). In the following year (15–
154 18 July 2010), N applications were made using dual labelled ¹⁵NH₄¹⁵NO₃ (99% labelled;
155 SerCon, Crewe, UK), applied at rates of 0 (controls with distilled water only, referred to as
156 Cw), 0.04, 0.4 g Nm⁻² yr⁻¹ and 1.2 g Nm⁻² yr⁻¹ (referred to as 0.04N, 0.4N and 1.2N,
157 respectively) with five replicated plots each, to quantify the fate of the pollutant N.

158 N treatments were applied using a watering can in 10 l of distilled water per plot
159 (adjusted to pH 4 with HNO₃, in line with the pH of extreme deposition events). These
160 treatments were applied over 4 days in 2010, with one-quarter of the total amount applied per
161 day to simulate wet deposition events previously observed (Hodson et al., 2005; Kühnel et
162 al., 2011). The lowest N addition (0.04N) simulated one of the previously recorded extreme
163 N deposition events where 0.04 g N m⁻² yr⁻¹ (~40% of the annual atmospheric N input) was
164 deposited with rainfall in less than a week (Hodson et al., 2005). The highest N treatment of
165 1.2N was undertaken to allow comparison with other past studies that simulated chronic
166 (rather than extreme) N deposition in the High Arctic (Baddeley et al., 1994; Woodin, 1997;
167 Gordon et al., 2001; Street et al., 2015).

168 Since the long-range transported air masses in the Arctic have been observed to
169 consist of nitrate or ammonium aerosol which could sometimes result in deposition
170 dominated by one or the other form of N (Dickerson, 1985; Hole et al., 2009; Aas et al.,

171 2011), a further five plots were split in two and used to determine the fate of separate NO_3^-
172 and NH_4^+ N depositions. One half of each split plot (1.5 m \times 0.75 m) received single labelled
173 $^{15}\text{NO}_3^-$ ($\text{Na}^{15}\text{NO}_3$ solution in 5 l water; hereafter referred to as ‘only $^{15}\text{NO}_3^-$ ’) and the other
174 half received single labelled $^{15}\text{NH}_4^+$ ($^{15}\text{NH}_4\text{Cl}$ solution; hereafter referred to as ‘only $^{15}\text{NH}_4^+$ ’) at a rate of 0.4 g N m^{-2} yr^{-1} . Identical N treatments were then repeated in 2011 from 1 to 4
175 July with non-labelled NH_4NO_3 , NaNO_3 and NH_4Cl .
176
177

178 **2.3. Fate of ^{15}N : field sampling**

179 Soil and plant samples were taken from each plot after 7 (25 July 2010), 21 (8 August
180 2010) and 388 (10 August 2011) days of the ^{15}N treatment. At each harvest in each plot, a
181 knife was used to cut a 10 \times 10 cm intact turf to a soil-depth of 10 cm. The knife was
182 thoroughly cleaned between each soil sampling to avoid ^{15}N contamination. Two extra soil
183 samples of 2 \times 2 cm of 10 cm depth were also taken from each plot for further estimation of
184 ^{15}N in the soil. Each sample was stored in separate plastic bags in a cold bag with ice packs
185 and transported to the NERC Arctic Research Station (Ny-Ålesund), and stored at 4 °C prior
186 to processing within 24 h.

187 Leachate samples were collected using two mini-rhizon soil samplers (10 cm length,
188 2.5 mm diameter; Van Walt, Surrey, U.K.) per plot installed below the main rooting zone and
189 just above the mineral soil layer. Samples were collected after 7, 21 days in the ^{15}N
190 application season, and after 347 days in the subsequent growing season (with rhizon
191 samplers for that collection installed on day 344). Samples were frozen and returned to the
192 UK for further analysis.

193

194 **2.4. Sample processing and analysis**

195

196 From each 10 × 10 cm plant-soil monolith, the above ground plant material was
197 separated from the soil and the soil organic and mineral layers were weighed and divided into
198 sub-samples for further analyses. The (very) few stones present in both horizons were
199 removed. Fresh soil sub-samples were used for quantification of soil moisture, pH, inorganic
200 N and microbial C and N content. Gravimetric soil moisture was determined by drying sub-
201 samples (10–20 g) in the oven at 60 °C for three days. These oven-dried sub-samples were
202 then returned to the UK for determination of total soil N and ¹⁵N. “Plant available” inorganic
203 N(NO₃⁻ and NH₄⁺) in soil was measured in fresh sub-samples by KCl extraction (Allen,
204 1989). These KCl extracts were frozen and returned to the UK for further analysis.

205 Microbial C and N extraction was measured by chloroform fumigation using 0.5 M
206 K₂SO₄ (Vance et al., 1987), determined from the difference in C and N released between
207 fumigated (~48 h with ethanol free chloroform) and non-fumigated soils. Fumigation extracts
208 were prepared for ¹⁵N analysis by the diffusion method (Brooks et al., 1989).

209 Above-ground plant material was oven-dried at 80 °C prior to separation into
210 different plant fractions and analyses of above-ground plant biomass, total N and ¹⁵N content.
211 Above-ground plant material was separated into bryophytes (mosses and liverworts), lichens,
212 *Salix polaris* (the dominant vascular plant), other vascular plants (largely *Saxifraga*, *Oxyria*
213 and *Polygonum* species) and litter (a mix of dead *Salix* leaves and stems, and dead mosses).
214 Below-ground stems of *Salix polaris* were removed from the 10 × 10 cm soil samples and
215 roots from a 5 × 5 cm section down the full soil depth. Samples were washed and dried in an
216 oven at 60 °C for three days for analyses of biomass, total N and ¹⁵N content.

217 Dried plant and soil samples were ground to a fine powder and total N content and
218 ¹⁵N enrichment were determined by isotope ratio mass spectrometry (ANCA GSL 20-20,
219 PDZ Europa, Crewe, Cheshire, UK). Calculations of ¹⁵N recovered in plant and soil pools
220 were then determined using the standard equation (Powlson and Barraclough, 1993):

221
$$F = T (A_S - A_B) / A_F \quad (1)$$

222 where F is N recovered from the labelled addition (^{15}N (g)/g of sample), T is the total weight
223 of N in the sample (N (g)/g of sample), and A_S , A_B and A_F are the at.% ^{15}N in the treated
224 sample, unlabelled control and added label, respectively. The percentage of ^{15}N treatment that
225 was recovered in each pool was then calculated.

226 Total inorganic N concentrations of KCl extracts were determined by flow injection
227 analysis (FIAflow2, Burkard Scientific, Uxbridge, UK), using sodium salicylate and
228 sulphanilimide colorimetric reactions to determine soil NH_4^+ and NO_3^- content respectively.
229 C, N and ^{15}N content of microbial biomass was determined as the difference between
230 fumigated and non-fumigated soil extracts that were freeze-dried and subsequently analysed
231 using the isotope ratio mass spectrometer. To correct for incomplete extraction, a conversion
232 factor (KEC) of 0.35 was used for microbial C (Cheng and Virginia, 1993) and a factor of 0.4
233 for microbial N (KEN) (Jonasson et al., 1996).

234 Contribution of deposited N from extreme events to existing ecosystem N pools was
235 also measured by expressing ^{15}N as a percentage of the existing total N pool in each fraction.
236 All the above- and below-ground plant materials for *Salix* were added together for calculation
237 of total ^{15}N retention and contribution of ^{15}N to total N in the *Salix* pool.

238

239 **2.5. Statistical analyses**

240 Statistical analyses were carried out using SPSS 17.0 (SPSS Inc., Chicago, Illinois,
241 USA). Repeated-measures ANOVA (with date as the within-subject factor and treatment as
242 the between-subject factor) were used to determine overall N treatment effects and whether
243 there were significant changes in total ^{15}N recovery and ^{15}N retention in the different pools
244 with time. For each pool at individual samplings, data were compared using Tukey's HSD
245 tests to determine differences between treatment levels (0.04N, 0.4N and 1.2N) in ^{15}N

246 retention and ^{15}N proportion in the total N pool. A t-test was used to test differences between
247 $^{15}\text{NO}_3^-$ and $^{15}\text{NH}_4^+$ retention and their proportion of the total N for each pool at individual
248 sampling. Homogeneity of variances was tested with Levene's test and, where necessary, the
249 appropriate square root- or log-transformations were performed.

250

251 **3. Results**

252

253 **3.1. Recovery and partitioning of ^{15}N**

254 **3.1.1. Application of $^{15}\text{NH}_4^{15}\text{NO}_3$**

255 Total % recovery of ^{15}N was consistently high across all treatments with average total
256 recovery of $81\pm 3.9\%$, $96\pm 4.9\%$ and $92\pm 5.4\%$ at day 7, 21 and 388, respectively (days and
257 treatments not significantly different from each other) (Table B.1, Appendix B).

258 *Non-vascular plants:* Bryophytes were the largest single pool for ^{15}N , retaining
259 between 32 and 62% (Fig. 1a). Percentage ^{15}N retention showed a marginally significant
260 decline with increasing N input (repeated measures ANOVA: d.f. = 2, 12, $F = 3.42$, $P = 0.08$),
261 with % retention 30% smaller overall under the 1.2N treatment compared to 0.04N (repeated
262 measures, Tukey, $P < 0.05$). Retention of ^{15}N peaked at day 21 (retentions of 62, 54 and 43%
263 in the 0.04, 0.4 and 1.2N) and showed a decline (to 41, 35 and 32%) by day 388 (marginally
264 significant time effect, repeated measures ANOVA: d.f.= 2, 11, $F = 2.95$, $P = 0.08$).

265 Lichens (which constituted just 4% of plant biomass) immobilised 2–6% of the ^{15}N
266 (Fig. 1d). Unlike bryophytes, % retention of ^{15}N was relatively similar across N treatments
267 over the two growing season, with average retentions of 3.5, 3 and 3% under 0.04, 0.4 and
268 1.2N, respectively.

269

270 **Vascular plants:** *Salix polaris* (the dominant vascular plant) was the second greatest
271 pool for ^{15}N , accumulating between 5 and 30% of applied ^{15}N in its leaves, stems and roots
272 (Fig. 1b). This % retention remained relatively constant over the two growing seasons for all
273 the treatments. Unlike non-vascular plants, % retention of ^{15}N increased with the increasing
274 N input (repeated measures ANOVA: d.f. = 2, 12, $F = 11.43$, $P < 0.01$) with average retention
275 of ^{15}N being 6, 15 and 24% of the 0.04, 0.4 and 1.2N treatments respectively. Overall, % ^{15}N
276 retention under 1.2N treatment was 75% greater than that of the 0.04N treatment (repeated
277 measures, overall Tukey, $P < 0.01$). Other vascular plants which were a mix of *Saxifraga*,
278 *Oxyria* and *Polygonum* sp. Retained between 1 and 4% of the applied ^{15}N . These plants were
279 not widely present in all plots and there were no significant differences in % ^{15}N recovery
280 among the treatments (Fig. 1c).

281 **Litter:** The litter fraction represented the third greatest retention pool for ^{15}N ,
282 accumulating between 6 and 15%, irrespective of the treatment levels (Fig. 1e) with no
283 significant differences among the treatment levels.

284 **Organic Soil:** The organic soil fraction retained similar amounts of ^{15}N to the litter
285 fraction (4–15%). Retention was similar across N treatments with a modest increase in % ^{15}N
286 retention from day 21 to day 388 (not significant) (Fig. 1f).

287 Due to time constraints, microbial biomass could only be obtained for 0.4N and 1.2N
288 plots. At day 7, 70% of the ^{15}N in the organic soil was immobilised by the microbial biomass
289 irrespective of the treatment dose. This retention significantly decreased to 41 and 52% of the
290 soil ^{15}N by day 21 and day 388, respectively (time effect, repeated measures ANOVA: d.f. =
291 2, 11, $F = 7.11$, $P < 0.01$).

292 **Mineral Soil:** The mineral soil was the fifth greatest pool for ^{15}N , holding between 2
293 and 8% (Fig. 1g). By the end of the second season (day 388), ^{15}N % retention under 1.2N was

294 300 and 90% greater than under 0.04N and 0.4N (repeated measures ANOVA: d.f. = 2, 12, F
295 = 3.828, P < 0.05).

296 Of the ^{15}N in the mineral soil, 39% of this was immobilised by the microbial biomass
297 with no significant differences in the amount of ^{15}N recovered over time.

298 **Soil water:** The ^{15}N in soil water increased proportionally with the dose of N applied,
299 leaching 0.005, 0.5 and 4.8 mg $^{15}\text{N L}^{-1}$ soil water on day 7 (see Table 1 for Tukey's test)
300 which decreased by 75, 98 and 100% on day 347 under the 0.04, 0.4 and 1.2N treatments,
301 respectively (time* treatment interaction, repeated measures ANOVA, d.f. = 6, 30, F = 3.02,
302 P < 0.05) (Table 1). Loss of ^{15}N via leaching only represents <5% of the total ^{15}N applied
303 even if it is assumed that all applied water is leached.

304

305 **3.1.2. Separate application of $^{15}\text{NO}_3^-$ and $^{15}\text{NH}_4^+$**

306 Overall, both $^{15}\text{NO}_3^-$ and $^{15}\text{NH}_4^+$ were highly retained in the system. The main sinks
307 of NO_3^- were both bryophytes and *Salix polaris*, retaining between 33 and 49% and 31–39%
308 of N applied as $^{15}\text{NO}_3^-$, respectively while ^{15}N from $^{15}\text{NH}_4^+$ was mainly present in bryophytes
309 (35–65%). Differences between retention of ^{15}N from the separate additions of NO_3^- and
310 NH_4^+ for each fraction are described in detail in the Appendix B. In brief, bryophytes,
311 lichens, litter and microbial biomass showed no difference in overall retention of the ^{15}N
312 from NO_3^- and NH_4^+ (Fig. 2), except for bryophytes at day 7 with a significant greater
313 retention of ^{15}N from NH_4^+ (d.f.= 7, t = -5.06, P < 0.01). In contrast, there were large
314 significant differences in the ^{15}N retention of the two N forms in *Salix polaris* at all of the
315 sampling days, with much greater retention of ^{15}N from NO_3^- (31 to 39%) compared to ^{15}N
316 from NH_4^+ (6 to 8%) (repeated measures ANOVA: d.f. = 1, 8, F = 95.73, P < 0.001) (Fig.
317 2b). Organic and mineral soil also showed greater retention of $^{15}\text{NO}_3^-$ compared to $^{15}\text{NH}_4^+$ for
318 all of the sampling days (repeated measures ANOVA: P < 0.05) (Fig. 2f & g). There was ~20

319 times more $^{15}\text{NO}_3^-$ (1 mg $^{15}\text{N L}^{-1}$ soil water) than $^{15}\text{NH}_4^+$ in the soil water by day 7 (repeated
320 measures ANOVA, d.f. = 2, 11, F = 6.19, P < 0.05; Tukey, P b 0.05), and this difference
321 decreased to 3 times more $^{15}\text{NO}_3^-$ (0.02 mg $^{15}\text{N L}^{-1}$ water) than $^{15}\text{NH}_4^+$ by day 347
322 (time*treatment, repeated measures ANOVA, d.f.=4, 22, F=4.06, P < 0.05) (Table 1).
323 Approximations suggest a maximum loss of 2% of the applied $^{15}\text{NO}_3^-$ and <0.1% of the
324 applied $^{15}\text{NH}_4^+$ at day 7.

325

326 **3.2. Pollutant ^{15}N enrichment of existing N pools**

327 **3.2.1. $^{15}\text{NH}_4^{15}\text{NO}_3$ enrichment**

328 Contribution of the ^{15}N to existing N pools showed very similar patterns across the
329 day 7, 21 and 388 harvests, so here we focus on day 388 data to describe the longest-term
330 enrichment (Fig. 3a).

331 The contribution of the applied ^{15}N to the total N pools of all the fractions increased
332 proportionally with the increasing level of N treatments except for *Salix polaris*, where the
333 contribution of applied ^{15}N under the 1.2N treatment was much (~85 times) greater than for
334 the 0.04N treatment (Tukey, P b 0.001). The contribution of the ^{15}N to existing N pools was
335 greatest in lichens N bryophytes N *Salix polaris* N microbial biomass N litter N organic soil
336 N mineral soil.

337 In the non-vascular plant pool, treatment ^{15}N made up 0.41, 4.3 and 9.3% of the total
338 bryophyte N pool under the 0.04N, 0.4N and 1.2N treatments respectively, (ANOVA: d.f.= 2,
339 12, F = 53.50, P < 0.001) with all treatment levels being significantly different from each
340 other (Tukey, P < 0.01 between 0.04N and 0.4N, P < 0.001 for all other pairwise
341 comparisons). In lichens, these contributions were slightly greater at 0.59, 5.7 and 11.5%
342 (ANOVA: d.f. = 2, 12, F = 345.18, P < 0.001), again with all treatment levels being
343 significantly different (Tukey, P < 0.001 among all).

344 In *Salix polaris*, the ^{15}N contribution was lower when compared to non-vascular
345 plants, comprising 0.05, 0.99 and 4.7% of the existing N pools under the 0.04N, 0.4N and
346 1.2N treatments, respectively (ANOVA: d.f.= 2, 12, $F = 84.01$, $P < 0.001$; all treatment levels
347 were significantly different from each other, Tukey, $P < 0.01$).

348 In litter ^{15}N contributed 0.08, 0.87 and 2.1% of the total litter N pool (ANOVA: d.f. =
349 2, 12, $F = 47.68$, $P < 0.001$; all treatment levels were significantly different from each other,
350 Tukey, $P < 0.01$). While in the organic soil, these contributions were only 0.01, 0.07 and
351 0.32% (ANOVA: d.f.= 2, 12, $F = 23.41$, $P < 0.001$; Tukey, $P < 0.01$ for all comparisons), and
352 0.01×10^{-2} , 0.01 and 0.05% in the mineral soil, under the 0.04N, 0.4N and 1.2N treatments,
353 respectively (ANOVA: d.f.= 2, 12, $F = 24.76$, $P < 0.001$; Tukey, $P < 0.001$ between 0.04N
354 and 1.2N and between 0.4N and 1.2N).

355 In the microbial N pool, ^{15}N constituted 0.82 and 3.4% in the organic soil (d.f. = 8, $t =$
356 -2.940 , $P < 0.05$) and 0.15 and 0.69% in the mineral soil, under the 0.4N and 1.2N treatments
357 (d.f. = 8, $t = -2.88$, $P < 0.05$), respectively.

358

359 **3.2.2. Separate enrichment by $^{15}\text{NO}_3^-$ and $^{15}\text{NH}_4^+$**

360 At day 388, ^{15}N from NO_3^- and NH_4^+ contributed similar proportions to each of the N
361 pools with the exceptions of the vascular plant and soil pools (Fig. 3b). In the *Salix* pool, ^{15}N
362 from NO_3^- contributed significantly more (1.8%) of the N pool compared to ^{15}N from NH_4^+
363 (0.51%) (d.f.= 8, $t = 3.48$, $P < 0.01$) (Fig. 3b). In the organic soil horizon, ^{15}N from NO_3^-
364 contributed more to the existing N pool (0.14%) than N from NH_4^+ (0.06%), (d.f.= 8, $t =$
365 2.95 , $P < 0.05$) (Fig. 3b). There were similar differences in the mineral soil horizon, with ^{15}N
366 from NO_3^- contributing 0.04% compared to 0.002% from NH_4^+ (d.f. = 8, $t = 2.27$, $P = 0.05$)
367 (Fig. 3b).

368

369 **4. Discussions**

370

371 **4.1. Recovery of deposited $^{15}\text{NH}_4^{15}\text{NO}_3$ in plants, soil and microbial** 372 **fractions**

373 This study shows that High Arctic tundra has considerable capacity to accumulate the
374 pollutant N deposited in an extreme event, retaining ~95% across all treatments by the end of
375 the first growing season with ~90% still retained after the second growing season. This is of
376 further concern, given that tundra may be slow to recover from N deposition impacts (Street
377 et al., 2015). Other work using ^{15}N in arctic and alpine tundra habitats has also demonstrated
378 the ability of these ecosystems to act as rapid sinks for N (Bilbrough et al., 2000; Nordin et
379 al., 2004; Tye et al., 2005; Templer et al., 2012). Given the different aims and methods of ^{15}N
380 application in those studies, however, they cannot be used to estimate N accumulation from
381 summer extreme N deposition events (being performed either at snowmelt or using soil
382 injections, or not including doses or N recovery time-scales that relate to understanding
383 extreme N deposition impacts). Most relevant to our study was that of Tye et al. (2005),
384 which used snow-pack applied ^{15}N to trace its fate immediately after snowmelt. In that study,
385 much lower retention of ^{15}N was observed (~60% compared to 90% or greater in this current
386 study) which might indicate the importance of the timing of N deposition for N accumulation
387 (e.g. more active plant and microorganisms, soil thaw), although sites differences between
388 this and our study (e.g. plant cover, soil texture) may also explain the difference. The high
389 retention of N at our site was largely due to the ability of the bryophytes to retain incoming
390 NH_4NO_3 (Longton, 1997; Hyvarinen and Crittenden, 1998; Kotanen, 2002). This was likely
391 facilitated by their abundance, since they represented 47–61% of plant biomass and their high
392 N assimilation capacity (Choudhary, 2013). Our results also showed with increasing N dose,
393 the decreasing ^{15}N retention in the bryophytes was balanced largely by increasing ^{15}N

394 retention in *Salix polaris* and in both the soil horizons. This suggests that bryophytes might
395 reach their N-saturation capacity at deposition rates between 0.4 and 1.2 g N m⁻² yr⁻¹ and is
396 consistent with previous work which has reported N-saturation in mosses under 1 g Nm⁻² yr⁻¹
397 on Svalbard, as indicated by reduced nitrate reductase activity (Gordon et al., 2001).
398 Moreover, high retention after two growing seasons suggests N is absorbed by the bryophytes
399 rather than retained on the bryophytes surface.

400 There were some differences in short- (day 7 and 21) to medium term (day 388) ¹⁵N
401 retention in some of the pools but this did not affect the total recovery of ¹⁵N over time,
402 suggesting a tight ecosystem N cycling (Grogan et al., 2004). Microbial biomass in the
403 organic horizon showed the greatest changes in its ¹⁵N retention over time, retaining ~70% of
404 the ¹⁵N present in the soil during the initial partitioning (day 7) and decreasing later in the
405 season. This indicates a rapid microbial ¹⁵N turnover into both soil extractable and non-
406 extractable pools in the organic horizon (Tye et al., 2005; Clemmensen et al., 2008; Templer
407 et al., 2012). In contrast, the % ¹⁵N retention by the microbial biomass in the lower mineral
408 horizon that was able to immobilise ~30–40% of the ¹⁵N in the soil, remained unchanged
409 with time. This suggests a quick uptake of the ¹⁵N released from the dying microbial biomass
410 by new microbial biomass.

411 Furthermore, ¹⁵N analyses suggest in the organic soil horizon, the released ¹⁵N on
412 microbial turnover could be largely taken up by vascular plants, roots of which were mostly
413 found in that upper layer (Choudhary, 2013). *Salix*, thus appeared to have outcompeted
414 microbes for ¹⁵N after the initial partitioning (day 7), indicating its capacity to be an efficient
415 longer-term N sink together with the organic soil and bryophytes. In contrast, previous
416 studies have reported greater competitive ability of microbes compared to plants for nutrients
417 in arctic ecosystems (Nordin et al., 2004; Clemmensen et al., 2008). This may again be
418 attributed to the soil injection method (with N applied often below 10 cm) used in some past

419 work, which bypasses the moss layer and the main rooting zone that was the principal
420 location of N retention in our study. Our findings of surface applied N during the summer
421 months have greater parallels with the snowmelt period where there is transition from
422 microbial based N retention to plant-based retention by root uptake in the organic layer
423 (Brooks et al., 1998; Lipson and Monson, 1998). Microbial biomass in both the soil horizons,
424 thus, act as effective buffers for N leaching and help retain the deposited N in the soil by
425 either forming part of the extractable N pool for plants and microbes or the non-extractable N
426 pool that contributes to the total soil N pool.

427

428 **4.2. Importance of N form: fate of N from NO_3^- and NH_4^+**

429 Our results showed that bryophytes were the dominant sink for both the applied
430 $^{15}\text{NO}_3^-$ and $^{15}\text{NH}_4^+$. They showed only an initial preference for NH_4^+ , though this initial
431 preference could be the cause of their well-known sensitivity to reduced N (Mäkipää, 1995).
432 In the longer term they had no preference for either of the N forms. Many bryophyte species
433 have been previously reported to efficiently scavenge NH_4^+ (Yano et al., 2010) and NO_3^-
434 either by absorption across their entire surface (Turetsky, 2003) or through capillary
435 movement from the soil beneath (Press and Lee, 1982, Turetsky, 2003). Lichens and
436 microbial biomass also did not show any preference for $^{15}\text{NO}_3^-$ or $^{15}\text{NH}_4^+$ and previous work
437 also suggest that lichens are efficient nutrient immobiliser for NO_3^- and NH_4^+ (Tye et al.,
438 2005).

439 In contrast, vascular plants and especially *Salix*, generally competed better for
440 deposited NO_3^- than for NH_4^+ . This result agrees with earlier field studies of N uptake using
441 N-form mixtures in arctic ecosystems, which showed a greater uptake of NO_3^- than of NH_4^+
442 and amino acid N (Atkins et al., 1993; Nordin et al., 2004; Clemmensen et al., 2008; but see
443 McKane et al., 2002). The relatively higher mobility of NO_3^- in the soil, may also have

444 contributed to high plant access to NO_3^- in the rooting zone (data not shown) compared to
445 NH_4^+ , given that the diffusion rate of NO_3^- is fivefold higher than that of NH_4^+ (Jones et al.,
446 2005). Thus, deposition rich in nitrates could impact almost all the pools of this ecosystem
447 and may represent a major source of eutrophication.

448 We have not studied all form of N depositions, particularly organic N and particulate
449 matter. It is likely that these would show greater retention within the system, and through
450 microbial transformation end up in the same pools as the inorganic N that we have studied.
451 Empirical work is needed to test this.

452

453 **4.3. Pollutant N (^{15}N) contribution to existing N pools**

454 The % contribution of ^{15}N at all treatment levels to existing N pools was greatest in
455 the non-vascular plants, showing the capacity of such acute and low rates of N deposition to
456 enrich tissue N of these fractions. This also means non-vascular plants may be the first and
457 main component of this ecosystem to undergo measureable changes due to N enrichment
458 from extreme deposition events. This is consistent with other long-term studies that have
459 shown high sensitivity of bryophytes and lichens to N fertilisation (Gordon et al., 2001;
460 Madan et al., 2007).

461 ^{15}N contribution to the existing N pool of the vascular plants (mainly *Salix*) under the
462 1.2N treatment was ~85 times more than the 0.04N treatment (despite the lesser 30-fold
463 difference in dose between treatments). Similarly, the proportion of ^{15}N in the total microbial
464 N pool also increased with the increasing N load to a similar extent as seen for *Salix*. This
465 probably arises from the fact that most of the ^{15}N at the lowest N treatment is captured by the
466 non-vascular plants leaving less available for vascular plants or the microbial pool. At higher
467 N deposition rates, the non-vascular pool saturates and ^{15}N more readily breaks through to the
468 soil and rooting zone. None-the-less even the low ^{15}N treatments made modest contributions

469 to the vascular plant and microbial N pools, suggesting that multiple acute N deposition
470 events could gradually enrich these N pools.

471 High and equal contributions of ^{15}N from $^{15}\text{NO}_3^-$ and $^{15}\text{NH}_4^+$ into the non-vascular
472 plants- and microbial-N pools confirm that either of these N forms could enrich these pools
473 and may affect their structure and functions in a longer term. In contrast, the considerable
474 preference by *Salix* for NO_3^- over NH_4^+ resulted in ~4 times greater proportion of the applied
475 $^{15}\text{NO}_3^-$ than of the applied $^{15}\text{NH}_4^+$ in their total N pool. These findings are consistent with
476 Tye et al. (2005) who also reported a significantly greater proportion of ^{15}N in the *Salix* total
477 N pool from $^{15}\text{NO}_3^-$ compared with $^{15}\text{NH}_4^+$ in a snowmelt N application study.

478

479 **5. Conclusions**

480 This work shows that atmospheric N deposited in extreme events during summer
481 months is largely retained within the tundra regardless of the treatment rates and N forms,
482 and remains so over two growing seasons as a result of conservative N cycling. Although
483 there are differences in amount of N retained within different ecosystem compartments at
484 lower and higher N treatments, the loss of N from one pool (e.g. non-vascular plants) is
485 largely balanced by other pools (e.g. vascular plants and soil horizons). This also suggests
486 that non-vascular plants were important short-term sinks and were N saturated below 1.2 g N
487 $\text{m}^{-2} \text{yr}^{-1}$, while vascular plants and soil are important long term sinks. Microbial biomass in
488 both the soil horizons help retain the deposited N in the soil during the initial partitioning and
489 were, thus, effective buffers for N leaching. Between the two N forms, deposition rich in
490 ammonium would primarily affect non-vascular plants whereas nitrate could affect both non-
491 vascular and vascular plants. Substantial % contribution of ^{15}N to existing N pools suggested
492 that extreme deposition events may already be driving eutrophication of arctic tundra
493 ecosystems, which is of further concern given that these events are predicted to increase in

494 frequency in the future and that tundra may recover only slowly from N enrichment. In
495 conjunction with warming in this region, this may also have important implications for
496 primary productivity and hence the carbon balance of High Arctic tundra. Further studies are
497 needed to better understand long-term N retention as well as responses of plant and microbial
498 communities in tundra to such extreme N deposition events. Such studies can also improve
499 our understanding of critical N load of tundra ecosystem.

500

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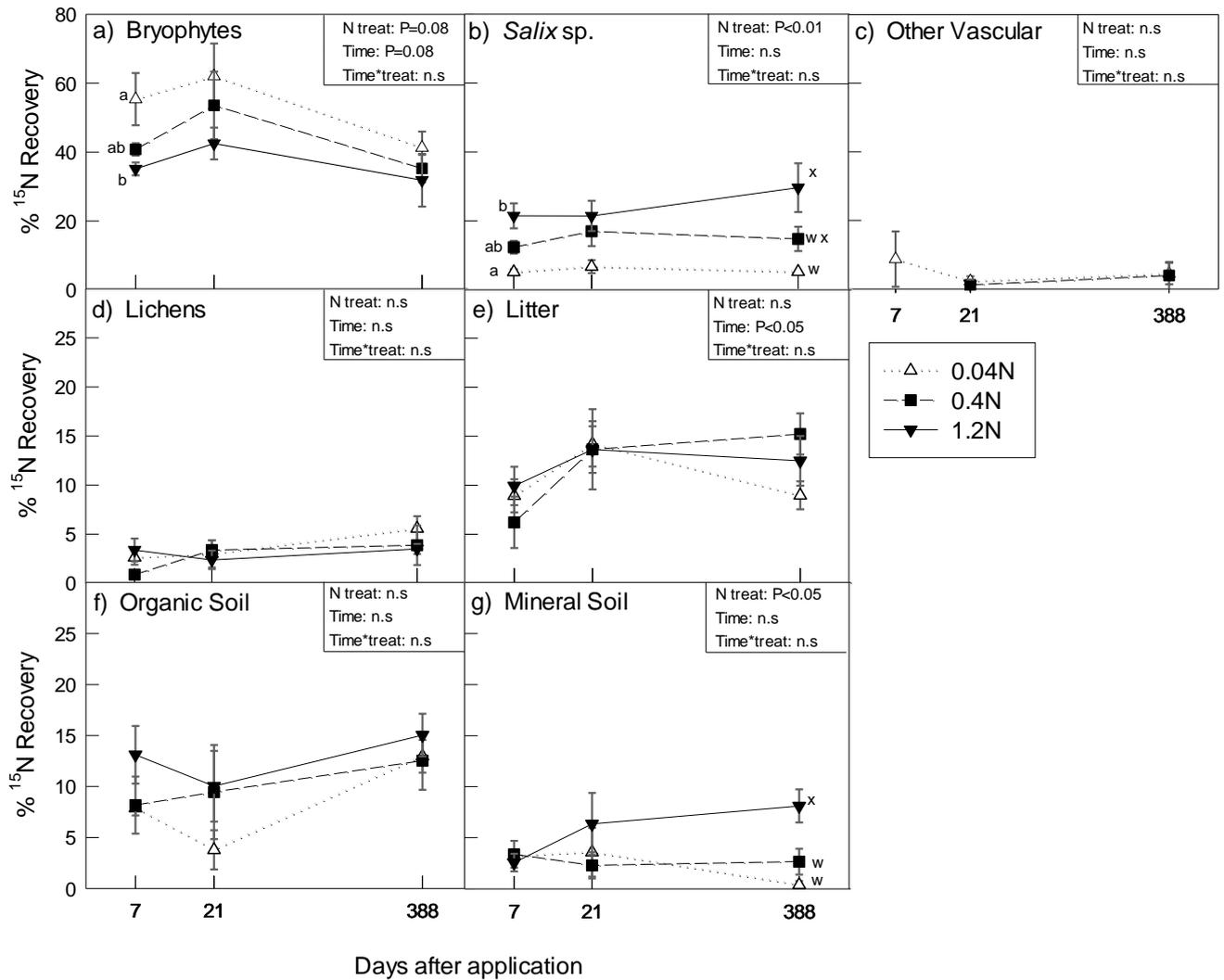
733 **Table 1.** Mg ¹⁵N per L soil water (means ± SE) from the plots treated with different doses
734 and forms of N after 7, 21 and 347 days of ¹⁵N application. Applications of ¹⁵NH₄¹⁵NO₃ at
735 rates of 0 g N m⁻²yr⁻¹ (Control), 0.04 g N m⁻²yr⁻¹ (0.04N), 0.4 g N m⁻²yr⁻¹ (0.4N), and 1.2 g
736 N m⁻²yr⁻¹ (1.2N), and Na¹⁵NO₃ (NO₃⁻) and ¹⁵NH₄Cl (NH₄⁺) at rates of 0.4 g N m⁻²yr⁻¹ rates
737 were applied in July 2010 and sampled after 7, 21 and 347 days of the treatment. Values
738 sharing the same letter within a date are not significantly different (Tukey HSD, P < 0.05). ±
739 are standard errors (N = 5).

740

Treatment	Days after ¹⁵ N application		
	7	21	347
Control	0.004 ± 0.001 ^a	0.004 ± 0.001	0.002 ± 0.0002
0.04N	0.005 ± 0.003 ^a	0.002 ± 0.0003	0.001 ± 0.0004
0.4N	0.48 ± 0.31 ^{ab}	0.08 ± 0.05	0.01 ± 0.007
1.2N	4.85 ± 2.23 ^b	3.22 ± 2.21	0.02 ± 0.01
NO ₃ ⁻	1.14 ± 0.57 ^b	0.19 ± 0.13	0.005 ± 0.002
NH ₄ ⁺	0.06 ± 0.05 ^a	0.006 ± 0.003	0.001 ± 0.0003

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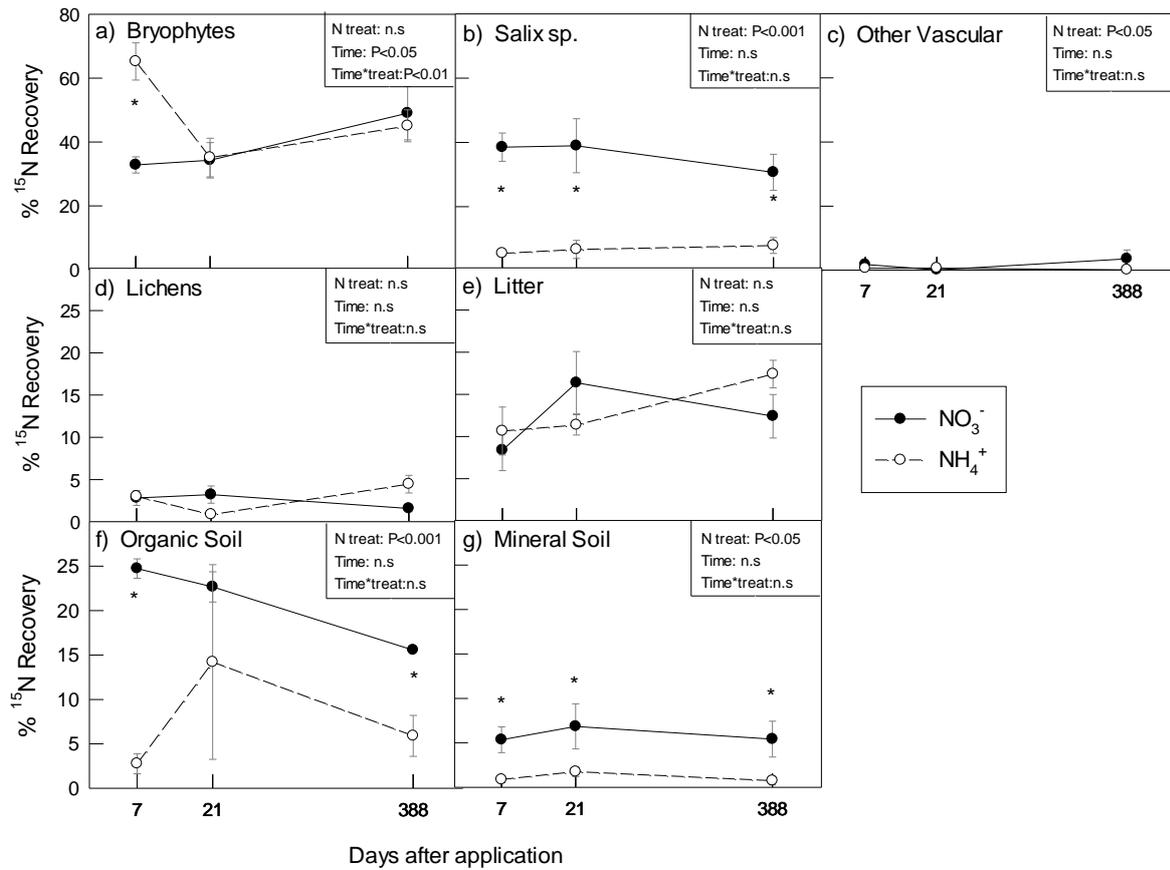


743

744 **Figure 1.** Mean % retention of applied ^{15}N (from $^{15}\text{NH}_4^{15}\text{NO}_3$) at 7, 21 and 388 days after
 745 treatment with applications of $0.04 \text{ g N m}^{-2}\text{yr}^{-1}$ (0.04N), $4 \text{ g N m}^{-2}\text{yr}^{-1}$ (0.4N), and $1.2 \text{ g N m}^{-2}\text{yr}^{-1}$ (1.2N). Dates of 7, 21 and 388 days correspond to the initial partitioning of ^{15}N , and
 746 recovery at the end of the first and second growing season respectively. Error bars represent
 747 one standard error and have been greyed for the clarity. Repeated measures ANOVA results
 748 shown for: N treat = N treatments, Time = time, Time*treatment = Time-treatment
 749 interaction; n.s = not significant. Different letters indicate significance within the same time
 750 point (Tukey HSD). Note the different scales of Y axes: 0-80 for a-c, 0-30 for d-g.

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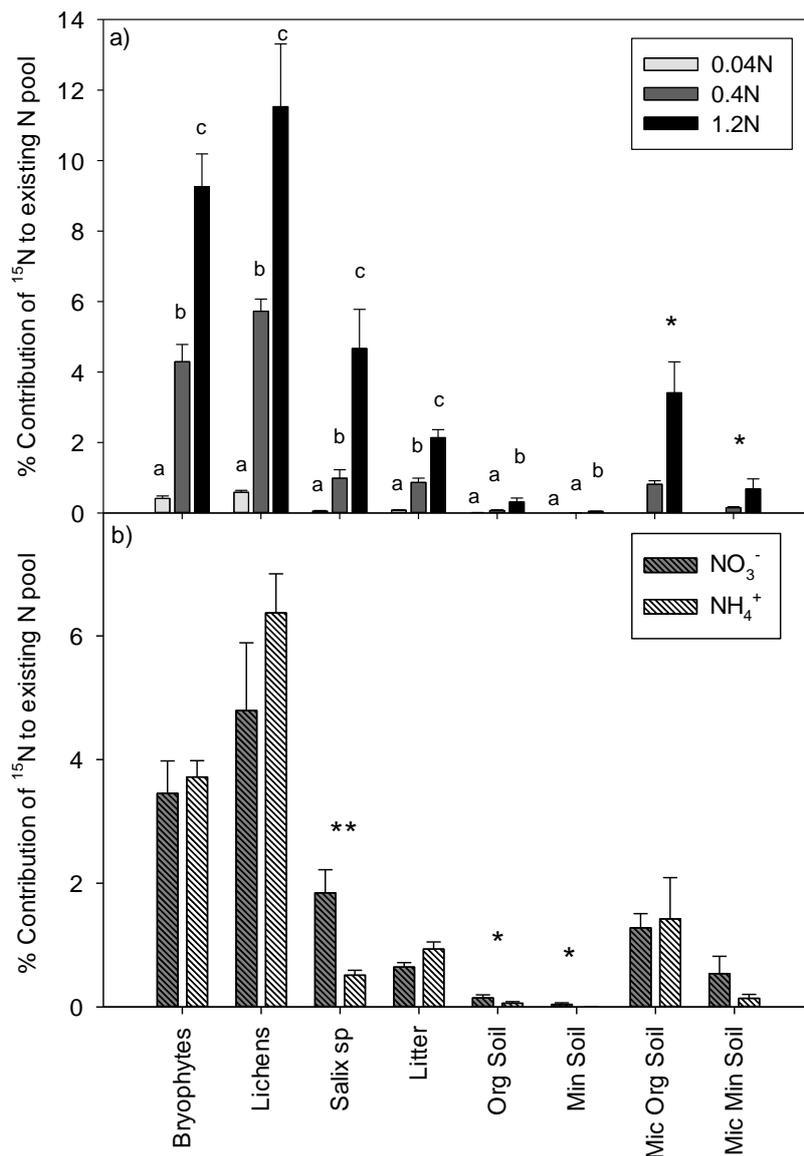
755 **Figure 2.** Mean % retention of applied ^{15}N from separate applications of $\text{Na}^{15}\text{NO}_3$ (NO_3^-)
 756 and $^{15}\text{NH}_4\text{Cl}$ (NH_4^+) at a rate of $0.4 \text{ g N m}^{-2}\text{yr}^{-1}$. Samples taken at 7, 21 and 388 days
 757 following treatment. Error bars represent one standard error (N = 5) and have been greyed for
 758 clarity. Repeated measures ANOVA results shown with abbreviations as for Fig 1. *
 759 indicates significant difference between $^{15}\text{NO}_3^-$ and $^{15}\text{NH}_4^+$ (t-test). Note the different scales
 760 of Y axes.

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766 **Figure 3.** Percentage ^{15}N contribution to the total N content of each pool after 388 days
 767 following application of (a) $^{15}\text{NH}_4^{15}\text{NO}_3$, and (b) $\text{Na}^{15}\text{NO}_3$ (NO_3^-) and $^{15}\text{NH}_4\text{Cl}$ (NH_4^+).
 768 Applications rates are for $^{15}\text{NH}_4^{15}\text{NO}_3$: $0.04 \text{ g N m}^{-2}\text{yr}^{-1}$ (0.04N), $0.4 \text{ g N m}^{-2}\text{yr}^{-1}$ (0.4N), and
 769 $1.2 \text{ g N m}^{-2}\text{yr}^{-1}$ (1.2N), and $0.4 \text{ g N m}^{-2}\text{yr}^{-1}$ for the separate $\text{Na}^{15}\text{NO}_3$ (NO_3^-) and $^{15}\text{NH}_4\text{Cl}$
 770 (NH_4^+) applications. Error bars represent one standard error ($N = 5$). Error bars in (a) that
 771 share the same letter are not significantly different (Tukey HSD, $P < 0.05$) and in (b)
 772 statistically significant differences shown as ** $P < 0.01$; * $P < 0.05$ (t-test). A t-test was
 773 performed for microbial fractions (Mic) in organic soil (Org soil) and mineral soil (Min soil)
 774 in graph (a) as there were just two treatments (0.4N and 1.2N) sampled.

775 **Appendix A**

776

777 **Supplementary material and methods**

778 **Experimental design**

779 In July 2009, 25 plots (1.5m × 1.5m) were established in approximately a 600 m² area
780 of tundra (Blaud *et al.*, 2015). On 21 and 22 August 2009, non-labelled NH₄NO₃ solution was
781 applied at rates of 0 (controls with distilled water only, referred to as “Cw”), 0.04 and 0.4 g N
782 m⁻² yr⁻¹ (“0.04N” and “0.4N”, five replicate plots each), with half of the total amount applied
783 per day. ¹⁵N applications were made in the following year (July 2010).

784 Precipitation and atmospheric N deposition data of Ny-Ålesund were obtained from
785 the Norwegian Institute for Air Research (NILU, Eklima, <http://ebas.nilu.no>) and air
786 temperature data were obtained from the Norwegian Meteorological Institute. In 2011, air
787 and soil temperatures were also measured at the experimental site throughout the growing
788 season using dataloggers (Tinytag Transit, Gemini, Chichester, UK).

789 Soil pH was determined in a 1:5 soil:water suspension.

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792 **Site N deposition, climate and vegetation**

793 Cumulative precipitation during the summer (July-August) was 17.5 mm and 57.7
794 mm in 2010 and 2011, respectively. Average inorganic N inputs during the summer
795 precipitation in 2010 and 2011 were low (0.0044 g NO₃⁻ m⁻² and 0.0040 g NH₄⁺ m⁻²,
796 respectively), with total annual N deposition of 0.064 and 0.112 g N m⁻² yr⁻¹ in those years,
797 respectively. Mean air temperatures during the growing season (July-August) at Ny-Ålesund
798 were 4.6 (±2.1) °C and 6.0 (±2.2) °C for 2010 and 2011, respectively. Mean pH of the organic
799 and mineral horizon was 6.68 and 7.19, respectively. Vegetation cover at our experimental
800 site was dominated by bryophytes (~40%), *Salix polaris* (~30%), lichens (~16%) and other

801 vascular plants. Further details on soils and vegetation N stocks are provided in Tables A.1 &
 802 A.2.

803

804 **Table A.1.** Soil pH, N and C content, and C/N ratios (means \pm SE) in control plots.
 805 Extractability factors of 0.40 for microbial N (N_{mic}) and 0.35 for microbial C (C_{mic}) were
 806 assumed. DW represents dry weight of the soil and BD is below detection limit. N_{total} is the
 807 total bulk soil N, N_{inorg} are the KCl extractable fractions, C_{total} is the total soil carbon; C_{CaCO_3}
 808 is the soil inorganic carbon.

Parameter	Organic Layer (O)	Mineral Layer (A)
Soil pH in H ₂ O	6.65 \pm 0.05	7.21 \pm 0.07
N_{total} (mg g ⁻¹ DW)	8.98 \pm 0.66	4.75 \pm 0.62
N_{inorg} (NO ₃ -N mg g ⁻¹ DW)	0.0007 \pm 0.0001	0.0005 \pm 0.0001
N_{inorg} (NH ₄ -N mg g ⁻¹ DW)	0.0073 \pm 0.0005	0.0025 \pm 0.0003
N_{mic} (mg g ⁻¹ DW)	0.48 \pm 0.09	0.06 \pm 0.01
C_{total} (mg g ⁻¹ DW)	168.09 \pm 9.76	49.21 \pm 6.00
C_{CaCO_3} (mg g ⁻¹ DW)	BD	BD
C_{mic} (mg g ⁻¹ DW)	3.67 \pm 0.62	1.56 \pm 0.28
C_{total} / N_{total}	20.10 \pm 1.91	10.54 \pm 0.28
C_{mic} / N_{mic}	7.15 \pm 0.70	18.54 \pm 3.28

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815 **Table A.2.** Mean dry matter and C and N stocks (g m^{-2}) of the soil and plant fractions.
 816 Values are means (\pm SE) of 20 plots and all samplings as there were no significant changes.

Fractions	Dry matter (g m^{-2})	N (g m^{-2})	C (g m^{-2})
Soil Organic Layer	6323 \pm 111	63 \pm 1.77	1011 \pm 26.39
Range	12287-4362	98-25	1993-492
Soil Mineral Layer	19788 \pm 559	113 \pm 5.38	1100 \pm 46.61
Range	32418-10046	230-31	2050-345
Bryophytes	549 \pm 27	4.07 \pm 0.20	198 \pm 10
Range	1554-112	10-0.98	581-37
Lichens	42 \pm 3.44	0.19 \pm 0.02	16 \pm 1.41
Range	160-1.70	0.72-0.01	83-0.60
<i>Salix</i> sp.	398 \pm 13	5 \pm 0.19	187 \pm 6.78
Range	737-184	11-1.72	379-83
Other vascular plants	69 \pm 13	0.79 \pm 0.13	29 \pm 5.82
Range	420-3	4-0.06	173-1.46
<i>Equisetum</i>	3 \pm 0.57	0.03 \pm 0.01	1.21 \pm 0.29
Range	6.3-0.20	0.07-0.01	3.47-0.11
Graminoids	9 \pm 5.70	0.07 \pm 0.04	3.22 \pm 2.11
Range	26-1.40	0.20-0.01	10-0.62
Litter	454 \pm 21	7 \pm 0.35	179 \pm 8.93
Range	1125-136	16-1.07	443-47

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819 **Appendix B**

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821 **Supplementary results & discussions**

822 **Fate of nitrate ($^{15}\text{NO}_3^-$) and ammonium ($^{15}\text{NH}_4^+$)**

823 Total recovery of ^{15}N from $^{15}\text{NO}_3^-$ in plant, soil and litter pools exceeded 100% (109
 824 to 115%) whereas recovery from $^{15}\text{NH}_4^+$ ranged between 71 and 96%. Overall $^{15}\text{NO}_3^-$ was
 825 taken up by all of the plant species (total “all-plant” fractions retained between 65% and 94%

826 of N from $^{15}\text{NO}_3^-$), while ^{15}N from $^{15}\text{NH}_4^+$ was mainly present in bryophytes (42-68%) with
827 total plant fractions retaining between 50 and 76% of N from $^{15}\text{NH}_4^+$.

828 **Non-vascular plants:** Bryophytes showed no difference in overall retention of the ^{15}N
829 from NO_3^- and NH_4^+ , retaining between 33 and 49% of N applied as $^{15}\text{NO}_3^-$, and 35 and 65%
830 applied as $^{15}\text{NH}_4^+$ (Fig. 2a). The only difference was a significant greater retention of ^{15}N
831 from NH_4^+ at day 7 (d.f=7, $t=-5.06$, $P<0.01$). Likewise, lichens did not show any differences
832 in ^{15}N retention from either form, retaining between 2 and 3% from NO_3^- and 1 and 5% from
833 NH_4^+ (Fig. 2d).

834 **Vascular plants:** In contrast, there were large significant differences in the ^{15}N
835 retention of the two N forms in *Salix polaris* at all of the sampling days, with much greater
836 retention of ^{15}N from NO_3^- (31 to 39%) compared to ^{15}N from NH_4^+ (6 to 8%) (repeated
837 measures ANOVA: $df=1,8$, $F=95.73$, $P<0.001$) (Fig. 2b). For all the other vascular plants
838 (Fig 2c), there were not enough plants present on the day 7 and 21 harvests for comparison
839 but at day 388, recovery from NO_3^- was ~5 times greater than from NH_4^+ (no statistical test
840 performed due to limited samples).

841 **Litter:** As with non-vascular plants, there were no differences between the retention
842 of N from $^{15}\text{NO}_3^-$ and $^{15}\text{NH}_4^+$, with an average retention of ~13% of the applied ^{15}N for either
843 form (Fig. 2e).

844 **Organic and mineral soil:** Organic soil showed greater retention of $^{15}\text{NO}_3^-$ (16 to
845 25%) compared to $^{15}\text{NH}_4^+$ (3 to 14%) for all of the sampling days (repeated measures
846 ANOVA: $d.f=1,8$, $F=155.57$, $P<0.001$) (Fig. 2f). Microbial biomass in the organic layer
847 showed no preference for either of the N forms applied and retained between 37 and 70% of
848 the ^{15}N pools in the organic soil applied in either form. As with the organic soil, the mineral
849 soil showed greater recovery of the applied ^{15}N from NO_3^- (5 to 7%) compared to ^{15}N from
850 NH_4^+ (1 to 2%) across all sampling days (repeated measures ANOVA: $d.f=1,8$, $F=8.29$,

851 P<0.05) (Fig. 2g). Microbial biomass in the mineral soil did not show any preference for
 852 either N form and immobilised between 16 and 51% of the ¹⁵N in the mineral soil pool.

853 **Soil water:** There was ~20 times more ¹⁵NO₃⁻ (1 mg ¹⁵N L⁻¹ soil water) than ¹⁵NH₄⁺
 854 in the soil water by day 7 (repeated measures ANOVA, d.f.=2,11, F=6.19, P<0.05; Tukey,
 855 P<0.05), and this difference decreased to 3 times more ¹⁵NO₃⁻ (0.02 mg ¹⁵N per L water) than
 856 ¹⁵NH₄⁺ by day 347 (time*treatment, repeated measures ANOVA, d.f.=4, 22, F=4.06, P<0.05)
 857 (Table 1). Approximations suggest a maximum loss of 2% of the applied ¹⁵NO₃⁻ and <0.1%
 858 of the applied ¹⁵NH₄⁺ at day 7.

859

860 **Table B.1.** Total % recovery of ¹⁵N (means ± SE). Applications of ¹⁵NH₄⁺/¹⁵NO₃⁻ at 0.04 g N m⁻²
 861 yr⁻¹ (0.04N), 0.4 g N m⁻²yr⁻¹ (0.4N), and 1.2 g N m⁻²yr⁻¹ (1.2N) rates, and Na¹⁵NO₃ (NO₃⁻)
 862 and ¹⁵NH₄Cl (NH₄⁺) at 0.4 g N m⁻²yr⁻¹ rates were applied from 15-19 July 2010 and sampled
 863 after 7, 21 and 388 days of the treatment. No statistically significant differences found
 864 between the treatments or within the days. ± are the standard errors (N = 5).

Treatments	D+7	D+21	D+388
0.04N	85 ± 7.6	95 ± 9.6	80 ± 4.7
0.4N	67 ± 2.2	100 ± 6.7	88 ± 3.2
1.2N	86 ± 6.8	98 ± 11.1	107 ± 13
NO₃⁻	109 ± 3.7	116 ± 5.1	111 ± 10.3
NH₄⁺	96 ± 6	71 ± 11.8	81 ± 3.3

865

866 **Total recovery of nitrate (¹⁵NO₃⁻) and ammonium (¹⁵NH₄⁺)**

867

868 Unexpectedly, total recovery of the applied ¹⁵NO₃⁻ (109-115%) was greater than that
 869 of ¹⁵NH₄⁺ (71-96%). However, a smaller loss of the more mobile NO₃⁻ seems unlikely since

870 lower concentrations of $^{15}\text{NH}_4^+$ than $^{15}\text{NO}_3^-$ were found in the leachate, and there would be
871 little NH_3 volatilization given the cool temperatures and wet conditions with rainfall
872 following the N applications (Nömmik & Vahtras 1982). Similar results were observed by
873 Templer *et al.*, (2012) in a meta-analysis of six tundra (combined arctic and alpine) sites.
874 However, the reasons for such higher recovery of NO_3^- compared to NH_4^+ , where soil
875 conditions do not support NH_3 volatilization, are still not well known. Nevertheless, given
876 total recovery of both N forms were high and not greatly different, deposition rich in either
877 forms have a capacity to affect different compartments of the tundra ecosystem. Deposition
878 rich in nitrates could impact almost all the compartments of this ecosystem whereas
879 ammonium deposition would primarily affect the non-vascular plants.